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## THE Be-Ta (BERYLLIUM-TANTALUM) SYSTEM

H. Okamoto  
L. E. Tanner

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Lawrence  
Livermore  
National  
Laboratory

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The Be-Ta (Beryllium-Tantalum) System  
9.01218 180.9479

By H. Okamoto and L.E. Tanner  
Lawrence Livermore National Laboratory

### Equilibrium Diagram

Six intermediate phases have been identified thus far, but little or no information is available concerning the alloy phase relations. Short reviews on this system have been provided by [66Sam] and [73Goll]. Figure 1 shows the position of each phase and melting points where available.

**( $\beta$ Be) and ( $\alpha$ Be) Terminal Solid Solutions.** The melting point of  $\beta$ Be and the  $\beta$ Be  $\rightarrow$   $\alpha$ Be allotropic transformation temperature are  $1289 \pm 4$  and  $1270 \pm 6$  °C, respectively [86BAP]. [50Kaul] observed a eutectic network in a slowly cooled alloy containing 0.05 at.% Ta. Hence, the solid solubility of Ta in Be is less than this value.

**Be<sub>12</sub>Ta (7.7 at.% Ta):** Mn<sub>12</sub>Th-type, identified by [57Bat] and [57Gla]. Its melting point is 1850 °C according to [59Pai, 60Sto]. The tetragonal compound found earlier by [36Mis] is likely to be this phase. There is no evidence of eutectic melting of Be-Ta compositions between Be<sub>12</sub>Ta and Ta below approximately 1700 °C [60Kri].

**Be<sub>17</sub>Ta<sub>2</sub> (10.5 at.% Ta):** Be<sub>17</sub>Nb<sub>2</sub>-type, identified by [59Pai(60Pai)] and [60Zal(61Zal)]. The melting point was reported as 1990 [59Pai], 1980 [60Sto], and 1988 °C [63Get].

**Be<sub>3</sub>Ta:** Be<sub>3</sub>Nb-type, identified by [60Zal(61Zal)]. Its melting point is not known.

**Be<sub>2</sub>Ta:** Cu<sub>2</sub>Mg-type with a melting point slightly below 1800 °C [59Pai]. This phase was also observed by [60Zal(61Zal)].

**Be<sub>2</sub>Ta<sub>3</sub>:** Si<sub>2</sub>U<sub>3</sub>-type, identified by [60Zal(61Zal)]. The melting point is not known.

**BeTa<sub>2</sub>:** Al<sub>2</sub>Cu-type, prepared by sintering at 1300 °C [65Gan]. Its melting point was not determined.

**(Ta) Terminal Solid Solution.** The melting point of Ta is 3020 °C [81BAP]. The initial slope of  $L/[L+(Ta)]$  liquidus in Fig. 1 is drawn by assuming no solubility of Be in (Ta) (see "Thermodynamics" section).

### Crystal Structures

A summary of crystal structure and lattice parameter data is given in Table 1. Thermal expansions of Be<sub>12</sub>Ta and Be<sub>17</sub>Ta<sub>2</sub> were measured by [61Boo] (Fig. 2 and Fig. 3) and are given in terms of averaged values by [65Bea] (Table 2).

[80Tan] predicted the existence of a stable or metastable compound, BeTa, having the CsCl-type crystal structure from the study of a series of Be-transition metal systems.

### Thermodynamics

Partial Gibbs energies for  $\text{Be}_{1/2}\text{Ta}$  and  $\text{Be}_{1/7}\text{Ta}_2$  at 1327 °C were calculated by [73Hull] (also in [73Spel]) from the results of vapor pressure measurements by [64Boo]:

Phase	$\bar{G}_{\text{Be}}$ , J/mol
$\text{Be}_{1/2}\text{Ta}$ (Ta-rich)	13707
$\text{Be}_{1/7}\text{Ta}_2$ (Be-rich)	13707
$\text{Be}_{1/7}\text{Ta}_2$ (Ta-rich)	22146

[74Zag] estimated the heat of mixing of the liquid Be-Ta as  $-6110X(1-X)$  J/mol using the method developed by [70Kau].

The information above is insufficient for phase diagram calculations.

The initial slope of the L/[L+(Ta)] liquidus has been calculated assuming that (1) the (Ta) phase has no homogeneity range and (2) the lattice stability parameter of solid Ta is  $-36570 + 11.105T$  J/mol; this was derived from the heat of fusion of Ta given by [83Chal].

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\* Indicates key paper.

Table 1 Be-Ta Crystal Structure and Lattice Parameter Data

Phase	Composition, at.% Ta	Struktur- Pearson bericht		Space group	Proto- type	Lattice parameters, nm		Reference
		symbol	designation			a	c	
( $\beta$ Be)....	0	cI2	A2	Im3m	W	0.25515	...	[82Kin]
( $\alpha$ Be)....	0	hP2	A3	P6 <sub>3</sub> /mmc	Mg	0.22857	0.35839	[81Kin]
Be <sub>1/2</sub> Ta...	7.7	tI26	D2 <sub>h</sub>	I4/mmm	Mn <sub>1/2</sub> Th	0.7334 0.7337 0.7330	0.4267 0.4255 0.4257	[57Bat] [57Gla] [58Che] (a)
Be <sub>1/7</sub> Ta <sub>2</sub> ...	10.5	hR19	...	R3m	Be <sub>1/7</sub> Nb <sub>2</sub>	0.739 0.7388	1.076 1.074	[59Pai] [60Zal] (b)
Be <sub>3</sub> Ta....	25	hR12	...	R3m	Be <sub>3</sub> Nb	0.453	2.095	[60Zal] (b)
Be <sub>2</sub> Ta....	33.3	cF24	C15	Fd3m	Cu <sub>2</sub> Mg	0.651 0.6507	... ...	[60Zal] (b) [75Stu]
Be <sub>2</sub> Ta <sub>3</sub> ....	60	tP10	D5 <sub>h</sub>	P4/mbm	Si <sub>2</sub> U <sub>3</sub>	0.650	0.332	[60Zal] (b)
BeTa <sub>2</sub> ....	66.7	tI12	C16	I4/mcm	Al <sub>2</sub> Cu	0.6009 0.6010	0.4902 0.4890	[65Gan] [72Hav]
(Ta).....	100	cI2	A2	Im3m	W	0.33031	...	[81Kin]

(a) Same group of authors as [57Bat]. (b) Also in [61Zal].

Table 2 Mean Linear Thermal Expansion Coefficients,  $\times 10^{-6} \text{m/m/}^{\circ}\text{C}$  [65Bea]

Compounds	Temperature range, $^{\circ}\text{C}$		
	27 to 316	27 to 982	27 to 1482
Be <sub>1/2</sub> Ta	10.4	13.7	15.1
Be <sub>1/7</sub> Ta <sub>2</sub>	11.2	14.0	15.7

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### Be-Ta Figure Captions

Fig. 1            Be-Ta Assessed Phase Diagram

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Fig. 2            Thermal Expansion of  $\text{Be}_{1/2}\text{Ta}$  (m/m)    [61Boo]  
                  ○ = Heating,        ● = Cooling.

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Fig. 3            Thermal Expansion of  $\text{Be}_{1/3}\text{Ta}_2$  (m/m)    [61Boo]  
                  ○ = Heating,        ● = Cooling.

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